

Reaction	Sources	Mean	SEM
Adenylate kinase	[1]	0.48	0.015
Phosphoglucose isomerase	[2, 3]	0.457	0.048
Aldolase	[2, 3]	0.084	0.017
Triose phosphate isomerase	[2, 3]	0.046	0.0023
Glyceraldehyde-3-phosphate dehydrogenase	[2, 3], NIST(50COR/VEL_252)*	0.066	0.017
Glycerol-3-phosphate dehydrogenase	NIST(58YOU/PAC_45+, 37EUL/ADL2_42+, 49BAR_43)	17085	1782
Phosphoglycerate kinase	[4], NIST(70KRI/BUC_579)	3377	88
Glycerol kinase	[5, 6]#, [7]	8.37e-4	4.8e-5
Phosphoglycerate mutase	[2, 3], NIST(49MEY/OES_1388, 59CHI/SUG_1391, 75GRI/CAR_1396)	0.17	0.0084
Enolase	[2, 4, 8], NIST(57WOL/BAL_1173)	4.17	0.75

**Table 1. Sources used for the calculation of the equilibrium constants mean and standard deviations.** The references retrieve from the NIST Standard Reference Database [9] are specified as NIST(id). SD=standard deviation of the values from all papers (for adenylate kinase the series of values given in [1] is used). \*=values corrected for pH as in [10]. +=values corrected for pH using a series of values at different pH. # =calculated from measured Km and Vmax using the Haldane equation.

## References

- Atkinson MR, Burton RM, Morton RK (1961) Equilibrium constant of phosphoryl transfer from adenosine triphosphate to galactose in the presence of galactokinase. Biochem J 78: 813–820.
- Voet D, Voet JG, Pratt CW (1999) Fundamentals of biochemistry. Wiley.
- Lehninger AL, Nelson DL, Cox MM (2005) Principles of biochemistry, volume 1. Worth.
- Bergmeyer HU (1974) Methods of enzymatic analysis. Verlag Chemie.
- Kralova I, Rigden DJ, Opperdoes FR, Michels PAM (2000) Glycerol kinase of *Trypanosoma brucei*. cloning, molecular characterization and mutagenesis. European Journal of Biochemistry 267: 2323–2333.
- Janson CA, Cleland WW (1974) The kinetic mechanism of glycerokinase. Journal of Biological Chemistry 249: 2562–2566.
- Burton K (1961) Biochemists' Handbook. In: Biochemists' Handbook, London: E. & F. N. Spon Ltd. C. long. edition, p. 94.
- Meyerhof O, Green H (1949) Synthetic action of phosphatase; equilibria of biological esters. The Journal of Biological Chemistry 178: 655–667.
- Goldberg RN, Tewari YB, Bhat TN (2004) Thermodynamics of enzyme-catalyzed reactions – a database for quantitative biochemistry. Bioinformatics 20: 2874–2877.
- Alberty RA (1998) Calculation of standard transformed Gibbs energies and standard transformed enthalpies of biochemical reactants. Archives of Biochemistry and Biophysics 353: 116–130.